Silicon Nitride-Silicon Carbide Nanocomposites Prepared by Water Processing of Commercially Available Powders

T. Kennedy,^{*a*} M. Poorteman,^{*b*} F. Cambier^{*b*} and S. Hampshire^{*a*}

^aCeramics Research Unit, University of Limerick, Limerick, Ireland b Belgian Ceramic Research Centre, 4, avenue Gouverneur Cornez, 7000 Mons, Belgium

Abstract

Dispersion of nano-sized silicon carbide powders in silicon nitride was examined. Zeta potential was measured over a range of pH values and the aqueous dispersion of nano-sized silicon carbide powders in silicon nitride matrix was optimised. Ultrafine silicon carbide, 30-2OOnm, which had agglomerated in organic media leading to a degradation in strength, was well dispersed and acted to reinforce silicon nitride matrix material. Denstfication of these nanocomposites was also examined. They were denstfied to 98-99% theoretical density by hot pressing, gas pressure sintering and pressureless sintering. Flexural strengths were measured and nanocomposites denstfied by the diflerent routes were compared. Gas pressure-sintered nanocomposites had strengths comparable to the hot pressed materials. Gas pressure sintering was identified as a viable densification route for these materials as it does not have the size and shape limitations associated with hot pressing. 0 1997 Elsevier Science Limited.

1 Introduction

Silicon nitride-silicon carbide composites have been prepared from silicon carbide fibres, platelets, whiskers and particulates. Although improvements in K_{IC} have been observed, these second-phase inclusions generally led to an overall degradation in strength. $1-3$

The emergence of superior silicon nitride-silicon carbide composites which contained nano-sized silicon carbide inclusions have renewed the efforts to develop materials with both improved strength and fracture toughness. These were fabricated by hot pressing amorphous Si-C-N powders formed from polymer precursors⁴ or by hot-pressing carbon-coated commercial $Si₃N₄$ powder.⁵ Detailed characterisation of these materials has been carried out and their properties are shown in Table 1. While these materials provide evidence to suggest that nano-sized silicon carbide reinforces silicon nitride to a very high degree, the mechanism by which it occurs is still very much in dispute and it has been suggested that the improvements in mechanical properties were as a result of very careful processing.8

The economic viability of these materials has also been questioned.⁹ Hot pressing is far from an ideal method of densification because of the limited shapes which can be produced. Ceramics formed from polymer precursors are very expensive and it is difficult to justify such a high cost raw material.

The present authors addressed some of these questions and attempted to produce $Si₃N₄-SiC$ nanocomposites using commercially available powders and following the established techniques for dispersion of micron-sized SIC inclusions as the second phase in $Si₃N₄$ matrix.^{9,10} Ball milling and attrition milling were used with isopropanol as vehicle to form slurries which were dried by evaporation. Silicon carbide agglomerates were a common feature of these composites and strength improvements were not observed.

It was obvious from this investigation that SiC powder dispersion was critical to the fabrication of these materials. The first advocates of nanocomposites, Niihara⁴ and Ishizaki,⁵ were not adding nano-sized SIC powder but, rather forming Sic particles during hot pressing, and thus agglomeration of SiC or dispersion in $Si₃N₄$ had not posed problems. Although some work has been carried out using a mixed powder route, $9-11$ as yet the $Si₃N₄$ materials originally prepared by Niihara and Ishizaki have not been reproduced by other laboratories.

| Niihara (1990/1991) ⁶ | | Ishizaki (1994) ⁷ | |
|--|--|---|--|
| Hot pressed $Si3N4-SiC$ nanocomposites from amorphous Si-C-N powders produced from precursors | | $Si3N4 – SiC nanocomposites by carbon$ coating and hot pressing | |
| Maximum values: K_{IC} : $\sigma_{\rm F\,r\,t}$: $\sigma_{\rm F~1200^{\circ}C}$ σ F 1500°C. | $7 \text{ MPa} \text{ m}^{0.5}$ 1350 MPa 1000 MPa 840 MPa | Maximum values: K_{IC} : $\sigma_{\rm F\,r\,t}$: $\sigma_{\rm F~1000^{\circ}C}$ $\sigma_{\text{F}~1200^{\circ} \text{C}}$ | 6.7 MPa.m ^{0.5} 1300 MPa 1230 MPa 1140 MPa |

Table 1. State of the art $Si₃N₄$ -SiC nanocomposites

Extensive investigations have been carried out on a powder-mixing route for the dispersion of Sic in \overline{Al}_2O_3 matrix.^{8,12,13} Notably, reproducible Al_2O_3 -SiC nanocomposites have been formed by processing with water and freeze-drying. Processing using water as opposed to organic solvents results in clear advantages for industrial-scale production of these materials.

The present work examines the possibility of a water-processing route for the preparation of $Si₃N₄ - SiC$ nanocomposites. It also compares hot pressing, gas pressure sintering and pressureless sintering of $Si₃N₄-SiC$ nanocomposites and compares the mechanical properties obtained for materials densified by the three different techniques.

2 **Experimental Procedure**

2.1 Materials used

Silicon nitride matrix material was prepared prior to addition of nano-sized silicon carbide powders. The processing of this matrix had previously been optimised and used as part of extensive investigations into $Si₃N₄-SiC$ composite materials by pressureless sintering.¹⁴ Si₃N₄ (H. C. Starck, Germany, LC12SX) and sintering additives (5.5 wt% Y_2O_3 , H. C. Starck, Grade C and 3 wt% Al_2O_3 Taimei, Japan, Taimicron TM-DR) were ball milled for 72 h, dried by hot plate-magnetic stirrer and the dried powder mix was brushed through a $212 \mu m$ sieve. Relatively high flexural strengths have been obtained for this matrix densified by pressureless sintering $(> 650 \text{ MPa})^3$ and it was therefore used as a starting point for $Si₃N₄-SiC$ nanocomposite production.

2.2 **Commercial nano-sized silicon carbide powders**

Two commercially available silicon carbide powders were mixed with pre-prepared silicon nitride matrix material. Detailed characterisation of Ibiden β -SiC-Betarundum, and Norton α -SiC-FCP40 was carried out and the average grain size, specific surface area and density of these powders are given in Table 2. Of the two powders, the

Norton has the lower average particle size and higher specific surface area. The density of the powders was the same at 3.15 g cm^{-3} .

2.3 **Dispersion**

The zeta potential of the commercial SiC powders, $Si₃N₄$ powder and $Si₃N₄$ pre-prepared matrix was measured for a range of pH values using a Coulter Delsa 440. Based on these measurements and on processing trials, the processing route outlined in the flowchart shown in Fig. 1 was established to fabricate reproducible $Si₃N₄-SiC$ nanocomposite powders.

Appropriate amounts $(5-10 \text{ vol})$ of nano-sized SiC powder were dispersed ultrasonically in water/ ammonia solution at $pH \sim 11$ for 1h and then milled for 1 h in an attrition mill with silicon nitride milling media, spindle and pot. $Si₃N₄$ pre-prepared matrix was added to the attrition mill and the slurry was milled for a further 3 h. The mill was kept at $\leq 20^{\circ}$ C to ensure pH \sim 11 was maintained throughout the milling procedure.

After milling, the aqueous slurry was freeze-dried (Finn Aqua Lyovac GT2 Freeze Dryer, France). Freeze-drying is a process which removes water from frozen material. Ice sublimes directly to water vapour avoiding the liquid state. This takes place under vacuum where the product is generally frozen between -10 and -70° C. Water vapour is extracted by way of a low temperature condenser. Pre-freezing should be at a speed of 0.1 to 2°C per min and in this case was carried out using liquid nitrogen.

2.4 **Densification**

The freeze-dried powder was densified by three different sintering routes:

- 1. Hot pressing at 1750°C for 1 h at 40 MPa in nitrogen.
- **2.** Pressureless sintering at 1850°C for 2 h in flowing nitrogen. Compacts for pressureless sintering were formed by CIP-ing the freezedried powders in latex moulds at 170 MPa. They were sintered in alumina crucibles in a tightly packed 50:50 $Si₃N₄:BN$ powder bed.

| Powder | Average grain size | <i>Specific</i> surface area BET/m^2g^{-1} | Density $\int g \, cm^{-3}$ gas Pycnometry | Crystal form |
|-------------------|-----------------------|--|---|-----------------|
| | [μ m] (50%) | | | |
| Norton FCP40 NLC | 0.20 | 30 | 3.15 | α |
| Ibiden-betarundum | 0.27 | 19.5 | 3.15 | |

Table 2. Characterisation of commercial silicon carbide powders

3. Gas pressure sintering (GPS) was also carried out on CIP-ed compacts in a similar powder bed to that used for pressureless sintering. A low pressure sintering step at 1800°C for 2 h was followed by a high pressure step at 1850 $^{\circ}$ C for 1 h. Nitrogen gas (0.3 MPa) and argon gas (0.3 MPa) were injected into the furnace chamber at room temperature before the GPS cycle had commenced. All further pressure increases during the GPS cycle were achieved using argon gas. The furnace was heated at 10° Cmin⁻¹ and pressure increased gradually with temperature until, at 1800°C a further injection of argon gas brought the pressure for the low pressure, sintering step to 2 MPa. After 2 h at 1800° C 2 MPa⁻¹, temperature and pressure were increased to 1850°C and 40 MPa, respectively, for 1 h.

Further details of this GPS/mixed gas procedure for successful densification of $Si₃N₄-SiC$ materials are detailed in previous publications by the authors.^{9,10}

Nanocomposites with 5 and 10 vol% nano-sized SiC were prepared. Pre-prepared matrix $Si₃N₄$ was milled in water for 3h without SiC, freeze-dried and sintered together with the composites.

Fig. 1. Water processing of $SiC-Si₃N₄$ nanocomposites.

2.5 **Characterisation**

The dense nanocomposite materials were characterised. Fired density was measured by upward displacement of water applying the principle of Archimedes. The phases present in all samples were identified using a Philips 'X'Pert' diffractometer (CuK_{α} radiation).

Test bars, $3 \times 4 \times 50$ mm were cut, groundpolished and chamfered, and flexural strength was estimated from 3-point bend tests (span, 16 mm; crosshead speed, 0.1 mm min^{-1}). Because of the importance and influence of the surface finish of test bars on measured strength, grinding and polishing was completely mechanised and a standard process used to prepare all bars tested.

3 **Results and Discussion**

3.1 Dispersion

Reproducible Al_2O_3-SiC nanocomposites have been formed by processing with water and freezedrying. O'Sullivan et al.¹³ dispersed SiC ultrasonically before adding to an $Al_2O_3/water$ slurry. A $pH \sim 6$ was maintained. At this pH, the Al₂O₃ particles have an overall positive charge and the SIC particles an overall negative charge. Sic repels SiC and attracts Al_2O_3 . The aqueous slurries were freeze-dried and Al_2O_3-SiC nanocomposites with strengths of ~ 800 MPa were formed by hot-pressing of the freeze-dried powder product.

Dispersion of SiC in $Si₃N₄$ differs from the Al_2O_3-SiC case in that both matrix and inclusion powders have a $SiO₂$ layer and effectively, dispersion involves mixing $SiO₂$ with $SiO₂$. The approach used for Al_2O_3-SiC is not directly applicable as both SiC and $Si₃N₄$ particles would have a similar overall charge at each pH.

The zeta potential of both commercial SIC powders, Ibiden β -SiC-Betarundum and Norton α -SiC FCP 40 was measured and is shown in Fig. 2. The carbide powders showed a minimum zeta potential at pH \approx 11. At this pH, the SiC particles have an overall negative charge and therefore repel each other.

Zeta potential data for silicon nitride powder and pre-prepared silicon nitride matrix are shown in Fig. 3. The sintering aids have little effect on

Fig. 2. Zeta potential of commercially available nano-sized silicon carbide powders.

Fig. 3. Zeta potential of LC12 SX Si₃N₄ powder as received and pre-prepared Si₃N₄ matrix (LC12SX + 3 wt% Al₂O₃ + 5.5 wt% Y_2O_3) ball milled in isopropanol.

powder behaviour as the zeta potential data points were similar with and without additives. As with SiC, minimum values for zeta potential were observed for $Si₃N₄$ at pH \approx 11. Because particles of both materials have an overall negative charge in aqueous media at this pH, particle agglomeration is prevented and a good dispersion of SIC in $Si₃N₄$ can be expected with intense mechanical agitation.

Drying of these aqueous slurries also needed careful consideration. Any heat or evaporation technique would inevitably lead to a change in the pH and therefore to the overall charge on the particles and their subsequent behaviour. For this reason, freeze-drying of the slurries was considered

to be the best option. The particles could be frozen into their well dispersed positions.

In addition, the resultant powders from the freeze-drying process are free-flowing and highly compressible, requiring no further treatment prior to compaction and sintering.

A processing route was thus established for the dispersion of nano-sized commercial Sic powders in $Si₃N₄$ matrix using water processing as outlined in the flowchart of Fig. 1.

3.2 Densification

Sintering was carried out by three different routes. Initially, the materials were hot-pressed to verify the good dispersion expected from the processing route used. Hot pressing at 1750°C for 1 h at α/β SiC were present in the nanocomposite speci-
40 MPa in nitrogen yielded dense composites with mens when densified by hot-pressing, gas pressure 40 MPa in nitrogen yielded dense composites with mens when densified by hot-pressing, gas pressure $\leq 0.1\%$ open porosity.

Pressureless sintering was carried out at a range of temperatures from 1700° C, but temperatures as high as 1850°C were necessary to obtain relative densities of $98-99\%$, comparable with those obtained through hot-pressing.

Gas pressure sintering was previously studied in detail for these materials but using a non-aqueous (organic liquid) processing route.^{9,10} GPS is considered an extremely viable densification alternative to hot pressing. It involves a two-step sintering cycle which can lead to the complete densification of pre-formed compacts but does not require glass or metal encapsulation. During the first step, a low pressure step, the compacts are sintered sufficiently to reduce porosity to $\lt 5\%$. At these porosity levels, higher pressures can be applied without the expensive and often unreliable encapsulation techniques necessary for HIP-ing. During the second high pressure step, the remaining porosity is eliminated.

Densification of $Si₃N₄-SiC$ composites by GPS poses a dilemma in that a nitrogen atmosphere is essential during sintering to prevent the decomposition of $Si₃N₄$. However, at certain nitrogen pressures and temperatures, the decomposition of silicon carbide under nitrogen occurs $3,9,15$ in the presence of a silica-rich liquid according to:

 $2SiC + SiO₂ + 2N₂ = Si₃N₄ + CO$ (1)

or

$$
3SiC + 3SiO2 + 2N2 = Si3N4 + 3SiO2 + 3CO (2)
$$

To retain the SiC without $Si₃N₄$ decomposition during GPS of these composites, a nitrogen/argon gas mixture was used and relative densities of 98 99% were obtained for $SiC-Si₃N₄$ nanocomposites with $\leq 0.1\%$ open porosity.

The presence of nano-sized silicon carbide did not inhibit densification. Unlike Si₃N₄-SiC composites with micron-sized inclusions,¹⁴ composites with ultrafine SiC, even up to 15 vol%, densified readily when conditions similar to those used to densify monolithic $Si₃N₄$ were employed.

3.3 **Characterisation**

X-ray diffraction analysis was carried out on all samples. The $\alpha \rightarrow \beta$ -Si₃N₄ transformation was complete in all cases. The added silicon carbide remained unchanged during densification and the Sic peaks grew in intensity as the volume percentage addition was increased. β -Si₃N₄ only was present in monolithic silicon nitride and β -Si₃N₄ with sintering or pressureless sintering.

SiC was well dispersed in the $Si₃N₄$ matrix. Figure 4 shows an SEM micrograph of a typical fracture face for these composites (5 vol% Ibiden β -SiC). Fracture sites were generally at surface flaws, usually at the chamfers which were not polished to the same mirror finish as the faces under tension. TEM confirmed that no Sic agglomeration had occurred and both intragranular and intergranular SiC was observed during preliminary TEM observations.

Figure 5 compares the strength of hot pressed and gas pressure sintered water processed monolithic $Si₃N₄$ with $Si₃N₄$ nanocomposites containing 5 vol% α - and β -SiC. The average strengths for monolithic hot-pressed and gas pressure sintered $Si₃N₄$ are 930 and 910 MPa, respectively. At 5 vol% SiC additions, significant improvements in strength were observed and strengths of >1 GPa were obtained for the nanocomposite materials. With either hot pressed or gas pressure sintered nanocomposites, higher average strengths were obtained for the nanocomposite with β -SiC inclusions.

The mechanical properties of pressureless sintered materials are generally lower than materials sintered with pressure. However, the same trends were evident as for pressure-assisted sintered nanocomposites. Inclusion of nano-sized Sic enhanced the mechanical properties of pressureless sintered $Si₃N₄$. The average flexural strength $(n = 7)$ of water processed monolithic $Si₃N₄$ pressureless sintered at 1850°C was 660 MPa. Strength for nanocomposite containing $5 \text{ vol} \%$ Norton FCP 40 was slightly higher at 680MPa but the average flexural strength of pressureless sintered nanocomposite with 5 vol% Ibiden Betarundum was 750 MPa. Again, β -SiC gave higher strengths than α -SiC.

Fig. 4. Fracture face of hot pressed $Si₃N₄$ 5 vol% Ibiden Betarundum SiC.

Fig. 5. Strength of hot pressed and gas pressure sintered $Si₃N₄$ -SiC nanocomposites.

 α -SiC is hexagonal which is the same crystallographic system as β -Si₃N₄, but β -SiC is cubic. If the SiC within β -Si₃N₄ grains is of a different crystal structure, then this may in some way influence the effect of the SIC particle within the grain. This is tentatively suggested at this stage and detailed TEM is required to investigate the effects of α - or β -SiC powder as reinforcement.

Increasing the volume fraction of silicon carbide led to further increases in the average strength. Figure 6 compares the strength of monolithic $Si₃N₄$ and nanocomposites containing 5 , 7.5 and 10 vol% Norton FCP 40 α -SiC, densified by hot pressing. The average strength increased from 940 MPa with 5 $vol\%$ SiC to 1110 MPa with 7.5 $vol\%$ SiC present. At 10 vol%, the average strength, 1090 MPa, was very slightly lower. However, there was a wider range of values for this composite and strengths of up to 1275MPa were obtained for individual specimens.

The optimum volume percentage addition of SIC is still not known. Nanocomposite with 15 vol% Norton SIC have been prepared and are currently being characterised. Materials containing higher volume percentage additions of Ibiden SIC are also under investigation.

Fig. 6. Flexural strength of $Si₃N₄$ and $Si₃N₄$ containing 5, 7.5 and 10 vol% Norton FCP 40 α -SiC.

It is evident from this work that commercial ultrafine Sic powders can be well dispersed in $Si₃N₄$ to enhance its mechanical performance and, therefore, exploitation of these materials may be feasible since high strengths achieved in hot pressed silicon nitride nanocomposites can be reproduced by gas pressure sintering which requires unencapsulated pre-formed compacts thus alleviating the drawbacks associated with hot pressing.

4 **Conclusions**

Homogeneous $Si₃N₄-SiC$ nanocomposites have been produced by water processing commercially available powders. It was found that addition of nano-sized Sic, even at 5 vol%, significantly improved the strength compared with monolithic silicon nitride. Near-complete densification has been achieved by pressureless sintering and gas pressure sintering, two economically viable sintering alternatives to hot-pressing and the high strengths achieved by hot pressing have been reproduced using gas pressure sintering.

Acknowledgements

The authors wish to acknowledge financial support from the European Commission (DGXII) for this work which was carried out as part of a joint Brite-Euram project (Contract No. BRE2-CT92-0358).

References

- 1. Lange, F. F., Effect of microstructure on strength of Si₃N₄-SiC composite system. *Journal of the American Ceramics Society, 1973, 56(9), 445-450.*
- *2.* Poorteman, M., Descamps, P., Sakaguchi, S., Cambier, F., Gilbart, E., Riley, F. L. and Brook, R. J., Mechanical

properties of Si₃N₄-SiC platelet composites. *Journal of the European Ceramic Society,* 1991, *8, 305-309.*

- *3.* Song, Y.-J., Gunay, V., O'Sullivan, D., Pomeroy, M. J. and Hampshire, S., Pressureless sintered $Si₃N₄-SiC$ composites. *Key Engineering Materials*, 1994, 89-91, 433-*438.*
- *4.* Izaki, K., Hakkei, K., Ando, K., Kawakami, T. and Niihara, K., Fabrication and mechanical properties of $Si₃N₄-SiC$ composites from fine, amorphous Si-C-N powder precursors. In *Ultrastructure Processing of Advanced Ceramics, Vo1.71,* ed. J. D. Mackenzie and D. R. Ulrich. Wiley, New York, 1988, pp. 891-900.
- 5. Watari, K., Ishizaki, K. and Kawamoto, M., Behaviour of carbon in hot pressed silicon nitride grain boundaries and its influence on mechanical properties. *Journal Ceram. Sot. Jpn. Inter. Ed.,* 1988, 96, 128-134.
- 6. Niihara, K., Izaki, K. and Nakahira, A., Nanocomposite materials based on Si₃N₄ and SiC. *Key Engineering Mate*rials, 1991, 56-57, 319-326.
- 7. Ishizaki, K. and Yanai, T., $Si₃N₄$ grain boundary reinforcement by Sic nanocomposites. *Silicates Industriels,* 1995, *7-8,215-222.*
- *8.* Zhao, J., Stearns, L. C., Harmer, M. P., Chan, H. M., Miller, G. A. and Cook, R. F., Mechanical behavior of alumina-silicon carbide 'nanocomposites'. *Journal of the American Ceramics Society,* 1993, *76(2), 503-510.*
- 9. Kennedy, T., O'Neil, J. P., Hampshire, S., Poorteman, M. and Cambier, F., Densification of silicon carbide-silicon nitride nanocomposites. Accepted for publication in *Silicates Industriels,* 1996, 9-10, 201-205.
- 10. Kennedy, T., Hampshire, S., Poorteman, M. and Cambier, F., Fabrication of silicon nitride-silicon carbide nanocomposite ceramics. *Key Engineering Materials*, 1995, 99-100,257-264.
- 11. Sasaki, G., Suganuma, K., Fujita, T., Hiraga, K. and Niihara, K., Interface structure of $Si₃N₄$ matrix composite with nano-meter scale SIC particles. *Mat. Res. Sot. Symp. Proc., Materials Research Society,* 1993, 287, 335-340.
- 12. Borsa, C. E., Jiao, S., Todd, R. I. and Brook, R. J., Processing and properties of Al_2O_3/SiC nanocomposite *Journal* **of** *Microscopy,* 1995, 177(3), 305-312.
- 13. O'Sullivan, D., Poorteman, M., Descamps, P., Cambier, F., Leriche, A. and Thierry, B., Optimisation of aluminasilicon carbide dispersions and the fabrication of nanocomposite ceramic materials. *Key Engineering Materials,* 1995, 99-100, 247-256.
- *14.* Song, Y.-J., Silicon nitride-silicon carbide particulate composite ceramics. PhD thesis, University of Limerick, Limerick, Ireland, 1994.
- 15. Greil, P., Petzow, G. and Tanaka, H., Sintering and HIPping of silicon nitride-silicon carbide composite materials. *Ceramics International,* 1987, 13, 19-25.